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Preliminary communication

Preparation and X-ray structure of the bis(pentacarbonyltungsten) complex of 1,3-diphosphaallene with η^1 -coordination mode

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Abstract

The bis(pentacarbonyltungsten) complex of 1,3-bis(2,4,6-tri-t-butylphenyl)-1,3-diphosphaallene with η^1 -coordination mode has been isolated and its crystal structure determined by X-ray analysis.

Sterically protected phosphorus-containing multiple-bonded compounds are currently of much interest. Several reports have been published on phosphacumulenes, including phosphaallenes [1-4] and 1,3-diphosphaallenes [5-7] carrying the 2,4,6-trit-butylphenyl group (abbreviated to Ar) as a kinetically protecting group. These phosphacumulenes contain low-coordinated phosphorus atom(s) (coordination number 2) and are interesting molecules because of their several coordination modes as ligand to transition metals [8,9].

We have reported several transition metal complexes of Group 6 with these phosphaallenes [10], and demonstrated that the coordination to the Group 6 metal occurs at the phosphorus atom in η^1 -fashion by taking into account the chemical shift correlation and phosphorus-tungsten coupling constants in solution ³¹P NMR. On the other hand, diphosphaallene 1 has been known to coordinate platinum and palladium as a π -type ligand [11,12].

We now report preliminary results on the preparation and X-ray analysis of the bis(pentacarbonyltungsten) complex with 1,3-bis(2,4,6-tri-t-butylphenyl)-1,3-diphosphaallene.



Scheme 1.

The phosphaallene 1 was allowed to react with an excess of (tetrahydrofuran)- $W(CO)_5$ to give the monocoordinated complex 2 [10] and the dicoordinated complex 3. These complexes were purified by gel permeation column chromatography.

The structure of 3 was unambiguously established by X-ray crystallographic analysis, whereas attempts to get suitable crystals for X-ray analysis of complex 2 have failed so far. Figure 1 shows an ORTEP drawing [13] of the molecular structure of 3, which depicts the first X-ray analysis of the η^1 -type doubly coordinated 1,3-diphosphaallene. Some selected bond distances and angles for 3 together with the data for 1 [7] are listed in Table 1. The diphosphaallene part in 3 has a similar conformation to non-coordinated 1. The P-W bond distances are 2.522(6) Å and 2.517(5) Å. The P=C bond lengths are 1.62(2) Å and 1.63(2) Å. The W, P, C1, and C2 atoms and W', P', C1', and C2' atoms are coplanar within 0.03 Å and 0.09 Å, respectively, and these planes make an angle of 68.6(5)°. The C2-C7 atoms and



Fig. 1. Molecular structure of the tungsten complex 3.

Table 1 Some selected bond distances (Å) and angles (°)

^a Data taken from Ref. 7. ^b Values calculated from atomic coordinates deposited at the Cambridge Crystallographic Data Centre.

C2'-C7' atoms are also coplanar within 0.03 Å and 0.05 Å, respectively, and the former and the latter rings make angles of 88(1)° and 90(1)° with the W, P, C1, C2 and W', P', C1', C2' planes, respectively. Dihedral angles $C2-P\cdots P'-C2'$ and $W-P\cdots P'-W'$ are 71.4(8)° and 66.2(4)°, respectively. In the uncoordinated diphosphaallene 1, the dihedral angle $C2-P\cdots P'-C2'$ is 83.0°, whereas in the complex 3 the angle is compressed by about 10°, probably because of steric congestion on coordination, while the π -system of the Ar rings and $p\pi$ of the P=C system are perpendicular. The crystal structure of 3 is quite different from that observed for bis(triethylphosphine)palladium complex [12], where the P=C bond that is η^2 -ligated to palladium is lengthened to 1.73(3) Å.

Experimental

Preparation of $[W(CO)_5]_2[ArP=C=PAr]$ (3). The starting material, bis(2,4,6-tri-t-butylphenyl)-1,3-diphosphallene (1) was prepared from diphosphirane, obtained from bis(2,4,6-tri-t-butylphenyl)diphosphene and dichlorocarbene, followed by ring opening with methyllithium, as described before [6]. The phosphaallene 1 (89.3 mg, 0.158 mmol) and hexacarbonyltungsten(0) (304.2 mg, 0.864 mmol) were dissolved in THF (3 ml) and were irradiated with a mercury lamp at 0 °C for 13 h to give the tungsten complexes of 1. These complexes were purified by recycling preparative HPLC over gel permeation columns (Japan Analytical Industry, Co., Ltd., JAIGEL H1 + H2) using THF as eluent to give 3 in 27% yield and 2 [10] in 23% yield. 3: m.p. 174.5-176.5 °C (decomposition), ³¹P {¹H} NMR: δ_P 144.3 accompanied by satellite peaks of an AA'X pattern with ¹ $J_{PW} = 285.0$ Hz, ² $J_{P-C-P} = 97.7$ Hz, and ³ $J_{PW} = 17.0$ Hz due to ¹⁸³W of 14.4% natural abundance.

X-Ray structure determination of 3. Compound 3 was recrystallized from pentane. C₄₇H₅₄O₁₀P₂W₂, M = 1208.59, monoclinic, space group $P2_1/a$, a 19.988(6), b 21.989(5), c 11.688(3) Å, $\beta = 98.25(2)$, U 5084(2) Å³, Z = 4, D_c 1.579 g cm⁻³. 6655 reflections with $2\theta \le 45^{\circ}$ were recorded on a four circle diffractometer using graphite-monochromated Mo- K_{α} radiation. Of these 3803 [with $I > 2\sigma(I)$] were judged observed. The structure was solved using MULTAN 80 [14]. At the last stage of the refinement, hydrogen atoms (calculated) were included in the structure factor calculations only. Full matrix least-squares refinement [15] with anisotropic temperature factors for nonhydrogen atoms converged to R = 0.075. The large thermal motion and/or partial disorder of t-butyl and carbonyl groups are the cause of low resolution of the X-ray data, resulting in large e.s.d.'s of the atomic coordinates. Atomic coordinates and thermal parameters and bond lengths and angles have been deposited at the Cambridge Crystallographic Data Centre.

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